

Tabulka termodynamických dějů pro m kg ideálního plynu

Děj	Izobarický	Izochorický	Izotermický	Adiabatický	Polytropický
Def.	$p_1 = p_2 = p$	$V_1 = V_2 = V$	$T_1 = T_2 = T$	$Q_{12} = 0$	$n \in (-\infty, +\infty)$
Vzt. pVT	$\frac{T_1}{V_1} = \frac{T_2}{V_2} = \frac{p}{m \cdot r}$	$\frac{T_1}{p_1} = \frac{T_2}{p_2} = \frac{V}{m \cdot r}$	$p_1 \cdot V_1 = p_2 \cdot V_2 = r \cdot T$	$\frac{T_2}{T_1} = \left(\frac{p_2}{p_1}\right)^{\frac{\kappa-1}{\kappa}} = \left(\frac{V_1}{V_2}\right)^{\kappa-1}$ $p_1 \cdot V_1^{\frac{\kappa-1}{\kappa}} = p_2 \cdot V_2^{\frac{\kappa-1}{\kappa}}$	$\frac{T_2}{T_1} = \left(\frac{p_2}{p_1}\right)^{\frac{n-1}{n}} = \left(\frac{V_1}{V_2}\right)^{n-1}$
ΔU	$\Delta U = U_2 - U_1 = c_p m (T_2 - T_1)$	$\Delta U = U_2 - U_1 = c_v m (T_2 - T_1)$	$\Delta U = 0$	$\Delta U = U_2 - U_1 = c_p m (T_2 - T_1)$	$\Delta U = U_2 - U_1 = c_p m (T_2 - T_1)$
ΔI	$\Delta I = I_2 - I_1 = c_p m (T_2 - T_1)$	$\Delta I = I_2 - I_1 = c_p m (T_2 - T_1)$	$\Delta I = 0$	$\Delta I = I_2 - I_1 = c_p m (T_2 - T_1)$	$\Delta I = I_2 - I_1 = c_p m (T_2 - T_1)$
ΔS	$\Delta S = c_p m \ln \frac{T_2}{T_1}$	$\Delta S = c_v m \ln \frac{T_2}{T_1}$	$\Delta S = r \cdot m \ln \frac{V_2}{V_1} = r \cdot m \ln \frac{p_1}{p_2}$	$\Delta S = \frac{Q_{12}}{T} = 0 \Rightarrow S_2 = S_1 = \text{konst.}$	$\Delta S = c_v \frac{n-\kappa}{n-1} m \ln \frac{T_2}{T_1}$
Q_{12}	$Q_{12} = I_2 - I_1 = c_p m (T_2 - T_1)$ $Q_{12} = \frac{1}{\kappa-1} p (V_2 - V_1)$	$Q_{12} = U_2 - U_1 = c_v m (T_2 - T_1)$ $Q_{12} = \frac{1}{\kappa-1} V (p_2 - p_1)$	$Q_{12} = m \cdot r \cdot T \cdot \ln \frac{p_1}{p_2} = p_1 V_1 \ln \frac{V_2}{V_1}$ $A_{12} = Q_{12}$	$Q_{12} = 0$	$Q_{12} = c_v \frac{n-\kappa}{n-1} m (T_2 - T_1) = c_n m (T_2 - T_1)$ $Q_{12} = \frac{\kappa-n}{\kappa-1} A_{12} = \frac{1}{n} \frac{\kappa-n}{\kappa-1} A_{12}$
A_{12}	$A_{12} = p (V_2 - V_1) = m \cdot r (T_2 - T_1)$ $A_{12} = \frac{\kappa-1}{\kappa} Q_{12}$	$A_{12} = 0$	$A_{12} = m \cdot r \cdot T \cdot \ln \frac{p_1}{p_2} = p_1 V_1 \ln \frac{V_2}{V_1}$ $A_{12} = Q_{12}$	$A_{12} = U_1 - U_2 = \frac{1}{\kappa-1} m (T_1 - T_2)$ $A_{12} = \frac{p_1 V_1 - p_2 V_2}{\kappa-1} = \frac{p_1 V_1}{\kappa-1} \left[1 - \left(\frac{p_2}{p_1}\right)^{\frac{\kappa-1}{\kappa}} \right]$	$A_{12} = m \frac{r}{n-1} (T_1 - T_2) = \frac{p_1 V_1 - p_2 V_2}{n-1}$ $A_{12} = \frac{p_1 V_1}{n-1} \left[1 - \left(\frac{p_2}{p_1}\right)^{\frac{n-1}{n}} \right]$
A_{112}	$A_{112} = 0$	$A_{112} = V (p_1 - p_2) = m r (T_1 - T_2)$	$A_{112} = m \cdot r \cdot T \cdot \ln \frac{p_1}{p_2} = p_1 V_1 \ln \frac{V_2}{V_1}$ $A_{112} = A_{12} = Q_{12}$	$A_{112} = I_1 - I_2 = \kappa \cdot A_{12}$ $A_{112} = \frac{\kappa}{\kappa-1} p_1 V_1 \left[1 - \left(\frac{p_2}{p_1}\right)^{\frac{\kappa-1}{\kappa}} \right]$	$A_{112} = m \frac{r}{n-1} (T_1 - T_2)$ $A_{112} = \frac{n}{n-1} p_1 V_1 \left[1 - \left(\frac{p_2}{p_1}\right)^{\frac{n-1}{n}} \right]$
Dia. p-V					
Dia. T-s					